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an atom? Why? From fireworks to stars, the color of light is useful in finding out what's in matter. The emission of light by hydrogen and other atoms has played a key role in understanding the electronic structure of atoms.

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in Model 2. a. Which color of light has the most energy? b. Which color of light has the least energy? 12. Does a gas discharge tube filled with boron emit the same wavelengths of light as a tube filled with hydrogen? Use evidence from Model 2 to support your answer. 13.

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Model 1 - White Light
Color Photon Energy (\times
 10^{-21}) (J) Wavelength
Range (nm) Speed

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sentences describes

the relationship

between wavelength

and energy of light?

(#6c) answer choices .

As wavelength gets

shorter the energy ...

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Nuclear Atom model to explain how light interacted with the electrons in an atom to produce spectral lines. Propose how this might be done and what the absorption spectra of hydrogen and boron would look like.

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