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Practice Test Modern Chemistry 91 Chapter Test Name Class Date Chapter Test B, continued 27. Distinguish between ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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Holt McDougal Modern Chemistry 2 Chapter Test Chapter Test A,
continued ______ 3. The periodic law states that a. no two
electrons with the same spin can be found in the same place in
an atom. b. the physical and chemical properties of the elements
are functions of their atomic number. c. wave patterns repeat at
regular intervals.

Assessment Chapter Test A - Wag & Paws

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Modern Chemistry Chapter 21 Test

Modern Chemistry 68 Chapter Test Name Class Date Chapter Test A, continued _____21. Metal X replaces the ions of metal Y from solution, but it cannot replace the ions of metal Z from

| solution. The order these metals should have in the activity series (from top to bottom) is a. x, y, z b. z, x, y c. y, x, z d. z, y, x22. |
|--|
| Assessment Chapter Test A - clarkchargers.org Modern Chemistry 86 Chapter Test Name Class Date Chapter Test A, continued13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity14. When there is a small decrease in temperature, the average kinetic energy of the particles |
| Assessment Chapter Test A - Ed W. Clark High School Chemistry Chapter 21 Nuclear Chemistry Test Review Modern Chemistry 175 Nuclearchemistry CHAPTER 21 REVIEW Nuclear Chemistry SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. Match each of the following statements with the process (es) to which they apply, using one of the choices below: (1) fission only (3 |
| Chapter 21 Review Nuclear Chemistry Answers Modern Modern Chemistry 4 Chapter Test Name Class Date Chapter Test A, continued20. In general, which properties are common to elements within a single group on the periodic table? a. atomic number and atomic weight b. chemical properties and ability to bond to other atoms c. color and luster d. boiling point and freezing point21. |
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